

Surface Conversion of ZnO Tetrapods Produces Pinhole-Free ZIF-8 Layers for Selective and Sensitive H₂ Sensing Even in Pure Methane

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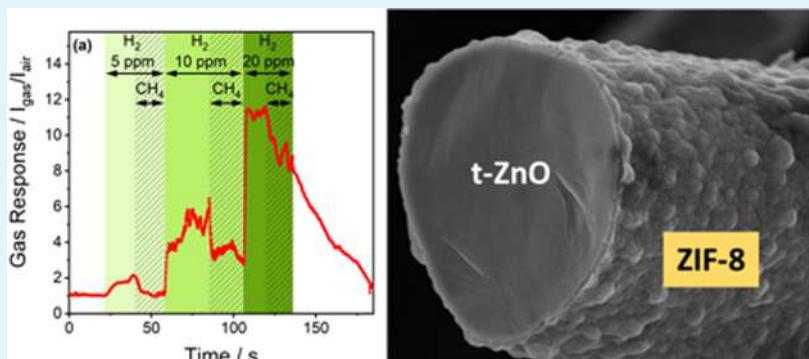
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ABSTRACT: As the necessary transition to a supply of renewable energy moves forward rapidly, hydrogen (H₂) becomes increasingly important as a green chemical energy carrier. The manifold applications associated with the use of hydrogen in the energy sector require sensor materials that can efficiently detect H₂ in small quantities and in gas mixtures. As a possible candidate, we here present a metal–organic framework (MOF, namely ZIF-8) functionalized metal-oxide gas sensor (MOS, namely ZnO). The gas sensor is based on single-crystalline tetrapodal ZnO (t-ZnO) microparticles, which are coated with a thin layer of ZIF-8 ([Zn(C₄H₅N₂)₂]) by a ZnO conversion reaction to obtain t-ZnO@ZIF-8 (core@shell) composites. The vapor-phase synthesis enables ZIF-8 thickness control as shown by powder X-ray diffraction, thermogravimetric analysis, and N₂ sorption measurements. Gas-sensing measurements of a single microrod of t-ZnO@ZIF-8 composite demonstrate the synergistic benefits of both MOS sensors and MOFs, resulting in an outstanding high selectivity, sensitivity ($S \cong 546$), and response times (1–2 s) to 100 ppm H₂ in the air at a low operation temperature of 100 °C. Under these conditions, no response to acetone, *n*-butanol, methane, ethanol, ammonia, 2-propanol, and carbon dioxide was observed. Thereby, the sensor is able to reliably detect H₂ in mixtures with air and even methane, with the latter being highly important for determining the H₂ dilution level in natural gas pipelines, which is of great importance to the energy sector.

KEYWORDS: zinc oxide tetrapods, metal–organic framework, ZIF-8, H₂ sensing, methane

INTRODUCTION

In view of the ongoing transition from fossil-based to renewable energy, zero-carbon energy sources, especially green H₂ are of particular interest as a replacement for natural gas. Therefore, in the near future, it is envisioned to use the already existing infrastructure and feed up to 20% H₂ into natural gas pipelines, as transport in tanks will not be profitable. In return, safety concerns must be considered, for example, leaks that require precise measurements as well as detection of small amounts of hydrogen along these pipelines. Thus, tiny and selective solid-state devices are required to be deployed along existing infrastructure. Several H₂ sensor technologies are known,^{1,2} and ongoing research focuses on higher selectivity and sensitivity,^{3–5} faster response, lower power consumption, and on meeting the demands of technical

applications.^{6,7} Around 20% of gas-sensing materials are based on metal-oxide materials, which operate in the presence of oxygen.⁸ One promising candidate is ZnO because of its good sensing response, easy fabrication, low cost, and chemical and thermal stability.³ ZnO can be made in many morphologies ranging from nanowires, nanorods, nanoparticles, and tetrapods (t-ZnO).⁹ The latter exhibits high crystallinity with low-defect density¹⁰ and provides high textural porosity for gases

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■ ABBREVIATIONS

ALD, atomic layer deposition
DMF, *N,N*-dimethylformamide
HMeIM, 2-methylimidazole
MOF, metal–organic framework
MOS, metal-oxide semiconductor
TGA, thermogravimetric analysis
VOC, volatile organic compound
(P)XRD, powder X-ray diffraction
FIB, focused ion beam

■ REFERENCES

- (1) Ren, Q.; Cao, Y.-Q.; Arulraj, D.; Liu, C.; Wu, D.; Li, W.-M.; Li, A.-D. Review—Resistive-Type Hydrogen Sensors Based on Zinc Oxide Nanostructures. *J. Electrochem. Soc.* **2020**, *167*, No. 067528.
- (2) Boon-Brett, L.; Bousek, J.; Black, G.; Moretto, P.; Castello, P.; Hübner, T.; Banach, U. Identifying Performance Gaps in Hydrogen Safety Sensor Technology for Automotive and Stationary Applications. *Int. J. Hydrogen Energy* **2010**, *35*, 373–384.
- (3) Bhati, V. S.; Hojaberdiel, M.; Kumar, M. Enhanced Sensing Performance of ZnO Nanostructures-Based Gas Sensors: A Review. *Energy Rep.* **2020**, *6*, 46–62.
- (4) Sowmya, B.; John, A.; Panda, P. K. A Review on Metal-Oxide Based p-n and n-n Heterostructured Nano-Materials for Gas Sensing Applications. *Sens. Int.* **2021**, *2*, No. 100085.
- (5) Kohlmann, N.; Hansen, L.; Lukan, C.; Schürmann, U.; Reimers, A.; Schütt, F.; Adelung, R.; Kersten, H.; Kienle, L. Fabrication of ZnO Nanobrushes by H₂–C₂H₂ Plasma Etching for H₂ Sensing Applications. *ACS Appl. Mater. Interfaces* **2021**, *13*, 61758–61769.
- (6) Hübner, T.; Boon-Brett, L.; Black, G.; Banach, U. Hydrogen Sensors – A Review. *Sens. Actuators, B* **2011**, *157*, 329–352.
- (7) Lukan, O.; Santos-Carballal, D.; Ababii, N.; Magariu, N.; Hansen, S.; Vahl, A.; Zimoch, L.; Hoppe, M.; Pauporté, T.; Galstyan, V.; Sontea, V.; Chow, L.; Faupel, F.; Adelung, R.; de Leeuw, N. H.; Comini, E. TiO₂/Cu₂O/CuO Multi-Nanolayers as Sensors for H₂ and Volatile Organic Compounds: An Experimental and Theoretical Investigation. *ACS Appl. Mater. Interfaces* **2021**, *13*, 32363–32380.
- (8) Drobek, M.; Kim, J.-H.; Bechelany, M.; Vallicari, C.; Julbe, A.; Kim, S. S. MOF-Based Membrane Encapsulated ZnO Nanowires for Enhanced Gas Sensor Selectivity. *ACS Appl. Mater. Interfaces* **2016**, *8*, 8323–8328.
- (9) Wang, Z. L. Nanostructures of Zinc Oxide. *Mater. Today* **2004**, *7*, 26–33.
- (10) Mishra, Y. K.; Modi, G.; Cretu, V.; Postica, V.; Lukan, O.; Reimer, T.; Paulowicz, I.; Hrkac, V.; Benecke, W.; Kienle, L.; Adelung, R. Direct Growth of Freestanding ZnO Tetrapod Networks for Multifunctional Applications in Photocatalysis, UV Photo-detection, and Gas Sensing. *ACS Appl. Mater. Interfaces* **2015**, *7*, 14303–14316.
- (11) Han, X.-G.; He, H.-Z.; Kuang, Q.; Zhou, X.; Zhang, X.-H.; Xu, T.; Xie, Z.-X.; Zheng, L.-S. Controlling Morphologies and Tuning the Related Properties of Nano/Microstructured ZnO Crystallites. *J. Phys. Chem. C* **2009**, *113*, 584–589.
- (12) Lukan, O.; Chow, L.; Chai, G. A Single ZnO Tetrapod-Based Sensor. *Sens. Actuators, B* **2009**, *141*, 511–517.
- (13) Li, B.; Wang, H.; Chen, B. Microporous Metal–Organic Frameworks for Gas Separation. *Chem. – Asian J.* **2014**, *9*, 1474–1498.
- (14) Li, J.-R.; Kuppler, R. J.; Zhou, H.-C. Selective Gas Adsorption and Separation in Metal–Organic Frameworks. *Chem. Soc. Rev.* **2009**, *38*, 1477–1504.
- (15) Koo, W.-T.; Jang, J.-S.; Kim, I.-D. Metal-Organic Frameworks for Chemiresistive Sensors. *Chem* **2019**, *5*, 1938–1963.
- (16) Ji, P.; Hu, X.; Tian, R.; Zheng, H.; Sun, J.; Zhang, W.; Peng, J. Atom-Economical Synthesis of ZnO@ZIF-8 Core–Shell Heterostructure by Dry Gel Conversion (DGC) Method for Enhanced H₂ Sensing Selectivity. *J. Mater. Chem. C* **2020**, *8*, 2927–2936.
- (17) Wu, X.; Xiong, S.; Mao, Z.; Hu, S.; Long, X. A Designed ZnO@ZIF-8 Core–Shell Nanorod Film as a Gas Sensor with Excellent Selectivity for H₂ over CO. *Chem. – Eur. J.* **2017**, *23*, 7969–7975.
- (18) Cui, F.; Chen, W.; Jin, L.; Zhang, H.; Jiang, Z.; Song, Z. Fabrication of ZIF-8 Encapsulated ZnO Microrods with Enhanced Sensing Properties for H₂ Detection. *J. Mater. Sci.: Mater. Electron.* **2018**, *29*, 19697–19709.
- (19) Lv, R.; Zhang, Q.; Wang, W.; Lin, Y.; Zhang, S. ZnO@ZIF-8 Core–Shell Structure Gas Sensors with Excellent Selectivity to H₂. *Sensors* **2021**, *21*, 4069.
- (20) Khudiar, A. I.; Elttayef, A. K.; Khalaf, M. K.; Oufi, A. M. Fabrication of ZnO@ZIF-8 Gas Sensors for Selective Gas Detection. *Mater. Res. Express* **2019**, *6*, No. 126450.
- (21) Park, K. S.; Ni, Z.; Côté, A. P.; Choi, J. Y.; Huang, R.; Uribe-Romo, F. J.; Chae, H. K.; O’Keeffe, M.; Yaghi, O. M. CCDC 602542: Exceptional Chemical and Thermal Stability of Zeolitic Imidazolate Frameworks. *Proc. Natl. Acad. Sci. U.S.A.* **2006**, *103*, 10186–10191.
- (22) Fairen-Jimenez, D.; Moggach, S. A.; Wharmby, M. T.; Wright, P. A.; Parsons, S.; Düren, T. Opening the Gate: Framework Flexibility in ZIF-8 Explored by Experiments and Simulations. *J. Am. Chem. Soc.* **2011**, *133*, 8900–8902.
- (23) Sutrisna, P. D.; Savitri, E.; Himma, N. F.; Prasetya, N.; Wenten, I. G. Current Perspectives and Mini Review on Zeolitic Imidazolate Framework-8 (ZIF-8) Membranes on Organic Substrates. *IOP Conf. Ser.: Mater. Sci. Eng.* **2019**, *703*, No. 012045.
- (24) Kim, H.; Kim, W.; Cho, S.; Park, J.; Jung, G. Y. Molecular Sieve Based on a PMMA/ZIF-8 Bilayer for a CO-Tolerable H₂ Sensor with Superior Sensing Performance. *ACS Appl. Mater. Interfaces* **2020**, *12*, 28616–28623.
- (25) Weber, M.; Kim, J.-H.; Lee, J.-H.; Kim, J.-Y.; Iatsunskyi, I.; Coy, E.; Drobek, M.; Julbe, A.; Bechelany, M.; Kim, S. S. High-Performance Nanowire Hydrogen Sensors by Exploiting the Synergistic Effect of Pd Nanoparticles and Metal–Organic Framework Membranes. *ACS Appl. Mater. Interfaces* **2018**, *10*, 34765–34773.
- (26) Kumar, A.; Mohammadi, M. M.; Zhao, Y.; Liu, Y.; Liu, J.; Thundat, T.; Swihart, M. T. Reduced Graphene Oxide-Wrapped Palladium Nanowires Coated with a Layer of Zeolitic Imidazolate Framework-8 for Hydrogen Sensing. *ACS Appl. Nano Mater.* **2021**, *4*, 8081–8093.
- (27) Koo, W.-T.; Qiao, S.; Ogata, A. F.; Jha, G.; Jang, J.-S.; Chen, V. T.; Kim, I.-D.; Penner, R. M. Accelerating Palladium Nanowire H₂ Sensors Using Engineered Nanofiltration. *ACS Nano* **2017**, *11*, 9276–9285.
- (28) Mishra, Y. K.; Kaps, S.; Schuchardt, A.; Paulowicz, I.; Jin, X.; Gedamu, D.; Freitag, S.; Claus, M.; Wille, S.; Kovalev, A.; Gorb, S. N.; Adelung, R. Fabrication of Macroscopically Flexible and Highly Porous 3D Semiconductor Networks from Interpenetrating Nanostructures by a Simple Flame Transport Approach. *Part. Part. Syst. Charact.* **2013**, *30*, 775–783.
- (29) Mishra, Y. K.; Kaps, S.; Schuchardt, A.; Paulowicz, I.; Jin, X.; Gedamu, D.; Wille, S.; Lukan, O.; Adelung, R. Versatile Fabrication of Complex Shaped Metal Oxide Nano-Microstructures and Their Interconnected Networks for Multifunctional Applications. *KONA Powder Part. J.* **2014**, *31*, 92.
- (30) Mishra, Y. K.; Adelung, R. ZnO Tetrapod Materials for Functional Applications. *Mater. Today* **2018**, *21*, 631–651.
- (31) Wang, B.-B.; Xie, J.-J.; Yuan, Q.; Zhao, Y.-P. Growth Mechanism and Joint Structure of ZnO Tetrapods. *J. Phys. D: Appl. Phys.* **2008**, *41*, No. 102005.
- (32) Ronning, C.; Shang, N. G.; Gerhards, I.; Hofäss, H.; Seibt, M. Nucleation Mechanism of the Seed of Tetrapod ZnO Nanostructures. *J. Appl. Phys.* **2005**, *98*, No. 034307.
- (33) Stassen, I.; Styles, M.; Grenci, G.; Gorp, H. V.; Vanderlinden, W.; Feyter, S. D.; Falcaro, P.; Vos, D. D.; Vereecken, P.; Ameloot, R. Chemical Vapour Deposition of Zeolitic Imidazolate Framework Thin Films. *Nat. Mater.* **2016**, *15*, 304–310.
- (34) Lukan, O.; Cretu, V.; Deng, M.; Gedamu, D.; Paulowicz, I.; Kaps, S.; Mishra, Y. K.; Polonskyi, O.; Zamponi, C.; Kienle, L.

- Trofim, V.; Tiginyanu, I.; Adelung, R. Versatile Growth of Freestanding Orthorhombic α -Molybdenum Trioxide Nano- and Microstructures by Rapid Thermal Processing for Gas Nanosensors. *J. Phys. Chem. C* **2014**, *118*, 15068–15078.
- (35) Lupon, O.; Chow, L.; Pauperté, T.; Ono, L. K.; Roldan Cuenya, B.; Chai, G. Highly Sensitive and Selective Hydrogen Single-Nanowire Nanosensor. *Sens. Actuators, B* **2012**, *173*, 772–780.
- (36) Lupon, C.; Khaledalidusti, R.; Mishra, A. K.; Postica, V.; Terasa, M.-I.; Magariu, N.; Pauperté, T.; Viana, B.; Drewes, J.; Vahl, A.; Faupel, F.; Adelung, R. Pd-Functionalized ZnO:Eu Columnar Films for Room-Temperature Hydrogen Gas Sensing: A Combined Experimental and Computational Approach. *ACS Appl. Mater. Interfaces* **2020**, *12*, 24951–24964.
- (37) Wu, H.; Zhou, W.; Yildirim, T. Methane Sorption in Nanoporous Metal–Organic Frameworks and First-Order Phase Transition of Confined Methane. *J. Phys. Chem. C* **2009**, *113*, 3029–3035.
- (38) Zhang, K.; Lively, R. P.; Dose, M. E.; Brown, A. J.; Zhang, C.; Chung, J.; Nair, S.; Koros, W. J.; Chance, R. R. Alcohol and Water Adsorption in Zeolitic Imidazolate Frameworks. *Chem. Commun.* **2013**, *49*, 3245–3247.
- (39) Demessence, A.; Boissière, C.; Gross, D.; Horcajada, P.; Serre, C.; Férey, G.; Soler-Illia, G. J. A. A.; Sanchez, C. Adsorption Properties in High Optical Quality NanoZIF-8 Thin Films with Tunable Thickness. *J. Mater. Chem.* **2010**, *20*, 7676–7681.
- (40) Ji, H.; Zeng, W.; Li, Y. Gas Sensing Mechanisms of Metal Oxide Semiconductors: A Focus Review. *Nanoscale* **2019**, *11*, 22664–22684.
- (41) Zhong, Y.; Djurišić, A. B.; Hsu, Y. F.; Wong, K. S.; Brauer, G.; Ling, C. C.; Chan, W. K. Exceptionally Long Exciton Photoluminescence Lifetime in ZnO Tetrapods. *J. Phys. Chem. C* **2008**, *112*, 16286–16295.
- (42) Postica, V.; Schütt, F.; Adelung, R.; Lupon, O. Schottky Diode Based on a Single Carbon–Nanotube–ZnO Hybrid Tetrapod for Selective Sensing Applications. *Adv. Mater. Interfaces* **2017**, *4*, No. 1700507.
- (43) Postica, V.; Gröttrup, J.; Adelung, R.; Lupon, O.; Mishra, A. K.; de Leeuw, N. H.; Ababii, N.; Carreira, J. F. C.; Rodrigues, J.; Sedrine, N. B.; Correia, M. R.; Monteiro, T.; Sonea, V.; Mishra, Y. K. Multifunctional Materials: A Case Study of the Effects of Metal Doping on ZnO Tetrapods with Bismuth and Tin Oxides. *Adv. Funct. Mater.* **2017**, *27*, No. 1604676.
- (44) Paulowicz, I.; Postica, V.; Lupon, O.; Wolff, N.; Shree, S.; Cojocaru, A.; Deng, M.; Mishra, Y. K.; Tiginyanu, I.; Kienle, L.; Adelung, R. Zinc Oxide Nanotetrapods with Four Different Arm Morphologies for Versatile Nanosensors. *Sens. Actuators, B* **2018**, *262*, 425–435.
- (45) Lupon, O.; Magariu, N.; Khaledalidusti, R.; Mishra, A. K.; Hansen, S.; Krüger, H.; Postica, V.; Heinrich, H.; Viana, B.; Ono, L. K.; Cuenya, B. R.; Chow, L.; Adelung, R.; Pauperté, T. Comparison of Thermal Annealing versus Hydrothermal Treatment Effects on the Detection Performances of ZnO Nanowires. *ACS Appl. Mater. Interfaces* **2021**, *13*, 10537–10552.
- (46) Lupon, O.; Ababii, N.; Santos-Carballeda, D.; Terasa, M.-I.; Magariu, N.; Zappa, D.; Comini, E.; Pauperté, T.; Siebert, L.; Faupel, F.; Vahl, A.; Hansen, S.; de Leeuw, N. H.; Adelung, R. Tailoring the Selectivity of Ultralow-Power Heterojunction Gas Sensors by Noble Metal Nanoparticle Functionalization. *Nano Energy* **2021**, *88*, No. 106241.
- (47) Fairen-Jimenez, D.; Galvelis, R.; Torrisi, A.; Gellan, A. D.; Wharmby, M. T.; Wright, P. A.; Mellot-Draznieks, C.; Düren, T. Flexibility and Swing Effect on the Adsorption of Energy-Related Gases on ZIF-8: Combined Experimental and Simulation Study. *Dalton Trans.* **2012**, *41*, 10752–10762.
- (48) An, G.; Xia, X.; Wu, S.; Liu, Z.; Wang, L.; Li, S. Metal–Organic Frameworks for Ammonia-Based Thermal Energy Storage. *Small* **2021**, *17*, No. 2102689.
- (49) Lupon, O.; Chai, G.; Chow, L. Fabrication of ZnO Nanorod-Based Hydrogen Gas Nanosensor. *Microelectron. J.* **2007**, *38*, 1211–1216.
- (50) Sharma, S. K.; Utpalla, P.; Bahadur, J.; Das, A.; Prakash, J.; Pujari, P. K. Crystal Size-Dependent Pore Architecture and Surface Chemical Characteristics of Desolvated ZIF-8 Investigated Using Positron Annihilation Spectroscopy. *J. Phys. Chem. C* **2020**, *124*, 25291–25298.
- (51) Zhang, C.; Gee, J. A.; Sholl, D. S.; Lively, R. P. Crystal-Size-Dependent Structural Transitions in Nanoporous Crystals: Adsorption-Induced Transitions in ZIF-8. *J. Phys. Chem. C* **2014**, *118*, 20727–20733.
- (52) Tanaka, S.; Fujita, K.; Miyake, Y.; Miyamoto, M.; Hasegawa, Y.; Makino, T.; Van der Perre, S.; Cousin Saint Remi, J.; Van Assche, T.; Baron, G. V.; Denayer, J. F. M. Adsorption and Diffusion Phenomena in Crystal Size Engineered ZIF-8 MOF. *J. Phys. Chem. C* **2015**, *119*, 28430–28439.
- (53) Pandey, R. K.; Dutta, J.; Brahma, S.; Rao, B.; Liu, C.-P. Review on ZnO-Based Piezotronics and Piezoelectric Nanogenerators: Aspects of Piezopotential and Screening Effect. *J. Phys. Mater.* **2021**, *4*, No. 044011.
- (54) Lupon, C.; Mishra, A. K.; Wolff, N.; Drewes, J.; Krüger, H.; Vahl, A.; Lupon, O.; Pauperté, T.; Viana, B.; Kienle, L.; Adelung, R.; de Leeuw, N. H.; Hansen, S. Nanosensors Based on a Single ZnO:Eu Nanowire for Hydrogen Gas Sensing. *ACS Appl. Mater. Interfaces* **2022**, *14*, 41196–41207.
- (55) Hobday, C. L.; Woodall, C. H.; Lennox, M. J.; Frost, M.; Kamenev, K.; Düren, T.; Morrison, C. A.; Moggach, S. A. Understanding the Adsorption Process in ZIF-8 Using High Pressure Crystallography and Computational Modelling. *Nat. Commun.* **2018**, *9*, No. 1429.
- (56) Kwon, H. T.; Jeong, H.-K.; Lee, A. S.; An, H. S.; Lee, J. S. CCDC 1429243: Experimental Crystal Structure Determination Cambridge Crystallographic Data Centre: U.K.; 2015.