

## Micro- and nano-structured oxides based on Cd- and Sn-doped InSe for volatile organic compound sensors applications

Sprincean Veaceslav <sup>1</sup>, Lupan Oleg <sup>2</sup>, Caraman Mihail <sup>1</sup>

<sup>1</sup> Moldova State University, e-mail: veaceslav.sprincean@usm.md, mihailcaraman@yahoo.com, ORCID: 0000-0001-6719-7387, 0000-0001-8225-8600

<sup>2</sup> Technical University of Moldova, e-mail: oleg.lupan@mib.utm.md, ORCID: 0000-0002-7913-9712

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**Abstract.** Indium oxide ( $\text{In}_2\text{O}_3$ ) is a material widely used in electronic devices as electrodes with high electrical conductivity and optical transparency [1]. The properties of this material extend with the transition from monocrystalline and polycrystalline thin layers to nanostructured films (nanowires, nanobelts, nanoparticles, etc.) [2]. In the nanostructured  $\text{In}_2\text{O}_3$  material, together with the characteristic properties of the  $\text{In}_2\text{O}_3$  compound, an intense luminescence in the visible range of the spectrum is emphasized, with maximum intensity in the blue region. In this work, micro and nanostructured  $\text{In}_2\text{O}_3$  and  $\text{In}_2\text{O}_3$ -InSe (Cd, Sn) layers and micro and nanostructures were fabricated by heat treatment (TT) in atmosphere of undoped InSe single-crystalline wafers doped with Cd and Sn in concentrations ranging from 0.25 at.% and 1.5 at.%. X-ray diffraction (XRD) studies, EDX plots and Raman spectra demonstrated that the layer formed on the surface of InSe (Cd, Sn) plates corresponds to the  $\text{In}_2\text{O}_3$  compound. Depending on the thermal annealing temperature, nanowires and nanobelts are obtained from  $\text{In}_2\text{O}_3$  microcrystallites with cubic and hexagonal lattice. The  $\text{In}_2\text{O}_3$  oxide formation process takes place in two steps. At a certain temperature the phase transformation  $\text{InSe} \rightarrow \text{In}_2\text{Se}_3$  and at the second stage  $\text{In}_2\text{Se}_3 \rightarrow \text{In}_2\text{O}_3$  takes place. The surface and edge defects of InSe lamellae serve as initiation centers for  $\text{In}_2\text{O}_3$  nanoforms. The forbidden band edge of the  $\text{In}_2\text{O}_3$  (Cd, Sn) layer is

formed by direct (3.6 eV) and indirect (~2.6 eV) optical transitions [3]. The nanostructured In<sub>2</sub>O<sub>3</sub> layers formed on the surface of InSe (Sn) and InSe (Cd) nanocrystals upon radiation excitation near the edge of the optical absorption band ( $\lambda=340$  nm) emit a photoluminescent radiation band (PL) covering the visible band (400÷700) nm range, with a well pronounced maximum in the 465÷475 nm region. The PL intensity in the maximum band as well as the PL band contour depend on several factors, such as excitation wavelength, excited beam intensity, temperature and time, and dopant concentration. In this paper, the influence of vapors of volatile organic compounds (benzol, acetone) on the peak intensity of the PL band was examined, from which it was determined that acetone vapors lead to the decrease of the PL intensity. The relaxation process of the PL intensity tends to saturation and is reached in ~25÷30 s and depends on the level of Cd and Sn doping of the primary InSe crystals from which the In<sub>2</sub>O<sub>3</sub> nanowire arrays were obtained. The ratio of the maximum PL band intensity of the In<sub>2</sub>O<sub>3</sub> nanowire layer in the initial not-annealed layer with vapor and its stationary PL intensity in vapor environment can serve as a criterion of the time and concentration of vapors, especially acetone vapors. This research was funded by the Ministry of Education and Research of the Republic of Moldova within the subprogramme no. 011210, "Advanced physical methods and UAV-based technologies for complex monitoring, evaluation and modeling" and the international project within the NATO Science for Peace and Security (SPS) program "Advanced technologies for physical resilience of critical infrastructures" (APRIORI), no. SPS MYP G6140.

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