

SENSING STUDIES OF COPPER OXIDE-ZINC OXIDE HETEROJUNCTIONS TO VOLATILE ORGANIC COMPOUNDS

Tudor ZADOROJNEAC ^{1*},
Nicolai ABABII ¹,
Nicolae MAGARIU ¹

¹ Technical University of Moldova, Faculty of Computers, Informatics and Microelectronics,
Department of Microelectronics and Biomedical Engineering, Center for Nanotechnology and Nanosensors,
PhD School Computer Science, Electronics and Energy, Chisinau, Republic of Moldova

* Corresponding author: Zadorojneac, Tudor, e-mail: zadorojneactudor@gmail.com

Abstract. *Detection and differentiation of volatile organic compounds (VOCs) is extremely important since presence of these gaseous pollutants in ambient air is harmful and poses a serious threat to human health even at small concentrations.*

In this paper, a simple and cost-effective method for synthesis of a nanostructured multilayer film of (CuO/Cu₂O)/ZnO:Fe is presented, which allows to form non-planar heterojunctions for efficient vapor detection of volatile organic compounds.

Keywords: *heterostructures, morphology, sensor, selectivity, sensitivity, VOCs*

Introduction

At the moment, there is a great interest for the research of volatile organic compounds sensors based on semiconductor oxides with improved and unique properties. In this context, the combination of two different materials, especially *n*- and *p*-types can cause electrical and chemical behavior specific to the goals, being very interesting and useful for applications in different fields, such as gas sensors with improved selectivity due to use of such heterostructures [1]. One of the most efficient combinations of metal oxides has been demonstrated for *n-p* heterostructures [2]. The most commonly used materials such as *n* and *p* type oxides are ZnO and CuO, respectively [3]. CuO/ZnO heterostructures are known as sensor structures, since they demonstrated a high selectivity to gases such as ethanol [4].

Other studies have shown that in the case of heterostructured nuclei, when the material interface participates in the detection mechanism, a higher gas response can be obtained due to the improved modulation of the electron depletion region [2]. Thus, the thickness of the upper layer of the structures must be comparable to the Debye length of the material [2]. The working function of the materials is another important parameter that must be considered, especially for the functionalized surface structures. As an example, Choi *et al.* have shown that local suppression/extension of the CuO nanowire hole accumulation layer can improve the gas detection properties, respectively the oxidation/reduction of the gases [5]. In this work, multilayered film of (CuO/Cu₂O)/ZnO:Fe that forms non-planar heterojunctions is studied for vapor detection of several volatile organic compounds.

Experimental

In this work, we report on the synthesis of Cu_xO/ZnO:Fe heterostructures produced by the SCS cost-effective chemical solution method [6]. Structural and morphological properties were investigated in detail. Gas detection studies have shown that heterostructures thermally annealed (TA) at 450 °C for 30 min have excellent selectivity to ethanol gas. This can be attributed to a decrease in the diameter of the nanocrystallites after thermal annealing, which leads to the formation of several potential barriers that affect the value of the response to the gas species, as well as from the oxidation of the Cu₂O phase to the crystalline CuO phase.

SEM analysis

Figure 1 shows the scanning electron microscopy (SEM) image of ZnO:Fe films with TA at 450°C for 30 min. The nanostructured film thickness is ~ 1.5 nm (measured in SEM section).

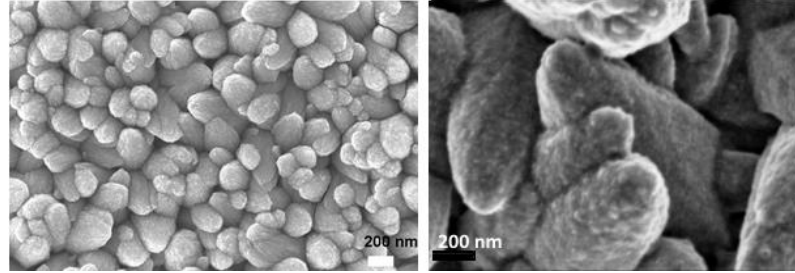


Figure 1. SEM image of CuO/Cu₂O/ZnO heterojunctions doped with 0.1 at% of Fe after thermal annealing at 450 °C for 30 min

The sample consists of column-type granules, densely packed with ZnO:Fe, which completely cover the glass substrate. No large clusters were observed, even on a large scale (see fig. 1). The diameter of the granules for the films of ZnO:Fe with about 0.1 at% of Fe varies in a wide range from 70 nm to 300 nm (see fig. 1).

In general, the sample is composed of granules with a rough surface which is very important for increasing the surface-volume ratio of the nanostructured films (see fig. 1). This is very attractive for detection applications due to its small size but with a larger contact surface [7]. Partial granular interconnectivity is another factor that enhances the detection properties of nanostructured films due to the specific mechanism described in a previous work for Fe doped ZnO [8].

XRD analysis

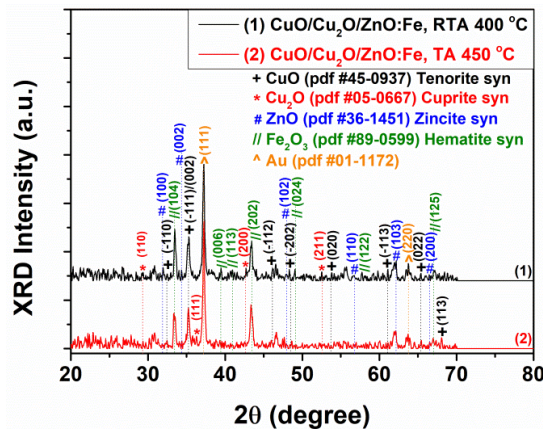


Figure 2. XRD patterns of CuO/Cu₂O/ZnO:Fe heterojunctions rapid thermal annealing (curve 1) and thermal annealing (curve 2)

The heterojunctions formation is demonstrated by the coexistence of the CuO, Cu₂O and ZnO phases shown in figure 2, such that at the values 2θ of 32.45°, 35.3°, 46.15°, 48.3°, 53.8°, 61.1°, 65.4° and 68.05° reflections (*hkl*) for CuO (Tenorite) copper oxide were obtained with Miller planes (-110) (-111)/(002), (-112), (-202), (020), (-113), (-220) and (113), respectively. The crystalline structure of Tenorite CuO is in monoclinic symmetry with space group *C2/c* [9] with constants $a=4.685 \text{ \AA}$, $b=3.425 \text{ \AA}$, $c=5.13 \text{ \AA}$, and $\beta=99.549^\circ$ [9-10].

Reflections at 2θ of 29.35°, 36.3°, 42.7° and 52.5° with Miller planes (110), (111), (200) and (211), respectively, it is assigned to the phase of Cu₂O (Cuprite) which has a cubic structure, of type, of the space group $T_h^2 - Pn3[65W]$ or $O_h^4 - Pn3[72P]$ [11]. Reflections at 2θ of 31.95°, 34.35°, 34.35°, 47.95°, 56.75°, 62.2° and 66.45° with Miller planes (100), (002), (102), (110), (103) and (200), respectively, it is attributed to the zinc oxide ZnO, according to the card (pdf # 36-1451) Zincite syn, which has a hexagonal structure [12].

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