



6th International Conference on Nanotechnologies and Biomedical Engineering  
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## A Nanosized Heteronuclear {Fe<sub>18</sub>Tb<sub>6</sub>} Coordination Wheel Based on Pivalate and Triethanolamine Ligands

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### Abstract

Ring-like systems assembled from metal ions and organic ligands attract attention due to variety of their useful properties and aesthetic beauty of their structure. A new nanosized heteronuclear [Fe<sub>18</sub>Tb<sub>6</sub>(piv)<sub>12</sub>(tea)<sub>6</sub>(Htea)<sub>18</sub>(N<sub>3</sub>)<sub>6</sub>]·n(solvent) (1) (where H<sub>2</sub>piv = pivalic acid) wheel-shaped cluster was synthesized by reacting of oxo-linked trinuclear Fe(III) pivalate precursors and terbium(III) nitrate with triethanolamine (H<sub>3</sub>tea) and sodium azide in the mixture of ethanol and acetonitrile under ultrasonic irradiation. Cluster 1 was characterized by elemental analysis and IR-spectroscopy. Single-crystal X-ray diffraction analysis shows that 18 iron(III) and 6 terbium(III) ions in 1 define a ring and are linked by 6 bridging pivalate and 24 triethanolamine ligands. Additionally, six pivalates and six azides completed the coordination sphere of the metal atoms in 1. In the resulting wheel core, three iron(III) and one terbium(III) ions generate six repeated {Fe<sub>3</sub>Tb} sequential fragments along the ring. The outer diameter of the {Fe<sub>18</sub>Tb<sub>6</sub>} wheel is ca. 3.5 nm, the inner diameter is ca. 1.0 nm, and the thickness of this molecular wheel is ca. 1.3 nm. In the crystal structure, the packing of bulky wheel-shaped clusters results in the formation of infinite channels fulfilled with solvent molecules. Upon removal of solvent molecules this structure reveals a huge total potential solvent accessible volume of ca. 26% per unit cell volume.

*Keywords:* iron, lanthanides, coordination clusters, X-ray structure

### References

1. Taft, K.L., Delfs, C., Papaefthymiou, G.C., Foner, S., Gatteschi, D., Lippard, S.J.: [Fe(OMe)<sub>2</sub>(O<sub>2</sub>CCH<sub>2</sub>Cl)]<sub>10</sub>, a molecular ferric wheel. *J. Am. Chem. Soc.* **116**(3), 823–832 (1994).

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2. Tasiopoulos, A.J., Vinslava, A., Wernsdorfer, W., Abboud, K.A., Christou, G.: Giant singlemolecule magnets: a Mn<sub>84</sub> torus and its supramolecular nanotubes. *Angew. Chem. Int. Ed.* **43**(16), 2117–2121 (2004). <https://doi.org/10.1002/anie.200353352>
3. Zhang, Z.M., Li, Y.G., Yao, S., Wang, E.B., Wang, Y.H., Clerac, R.: Enantiomerically pure chiral {Fe<sub>28</sub>} wheels. *Angew. Chem. Int. Ed.* **48**, 1581–1584 (2009). <https://doi.org/10.1002/anie.200805827>
4. Liu, C.-M., Zhang, D.-Q., Hao, X., Zhu, D.-B.: Nestlike C<sub>4</sub>-symmetric [Co<sub>24</sub>] metallamacrocycle sustained by p-tert-butylsulfonylcalix[4]arene and 1,2,4-triazole. *Chem. Eur. J.* **17**(44), 12285–12288 (2011). <https://doi.org/10.1002/chem.201101607>
5. Whitehead, G.F.S., Moro, F., Timco, G.A., Wernsdorfer, W., Teat, S.J., Winpenny, R.E.P.: A ring of rings and other multicomponent assemblies of cages. *Angew. Chem. Int. Ed.* **52**(38), 9932–9935 (2013). <https://doi.org/10.1002/anie.201304817>
6. Jiang, H., et al.: A gigantic molecular wheel of {Gd<sub>140</sub>}: a new member of the molecular wheel family. *J. Am. Chem. Soc.* **139**(50), 18178–18181 (2017). <https://doi.org/10.1021/jacs.7b11112>
7. Li, M., et al.: A family of 3d–4f octa-nuclear [Mn<sub>III</sub>4Ln<sub>III</sub>4] wheels (Ln = Sm, Gd, Tb, Dy, Ho, Er, and Y): synthesis, structure, and magnetism. *Inorg. Chem.* **49**(24), 11587–11594 (2010). <https://doi.org/10.1021/ic101754g>
8. Mondal, A., Raizada, M., Sahu, P.K., Konar, S.: A new family of Fe<sub>4</sub>Ln<sub>4</sub> (Ln = Dy<sub>III</sub>, Gd<sub>III</sub>, Y<sub>III</sub>) wheel type complexes with ferromagnetic interaction, magnetocaloric effect and zerofield SMM behavior. *Inorg. Chem. Front.* **8**(21), 4625–4633 (2021). <https://doi.org/10.1039/D1QI00781E>
9. Zou, L.-F., et al.: Adodecanuclear heterometallic dysprosium–cobalt wheel exhibiting singlemolecule magnet behaviour. *Chem. Commun.* **47**(30), 8659–8661 (2011). <https://doi.org/10.1039/C1CC12405F>
10. Baniodeh, A., et al.: Heterometallic 20-membered Fe<sub>16</sub>Ln<sub>4</sub> (Ln = Sm, Eu, Gd, Tb, Dy, Ho) metallo-ring aggregates. *Dalton Trans.* **40**(16), 4080–4086 (2011). <https://doi.org/10.1039/C0DT01742F>
11. Zhang, Z.-M., et al.: Wheel-shaped nanoscale 3d–4f Co<sub>II</sub>16Ln<sub>III</sub>24 clusters (Ln = Dy and Gd). *Chem. Commun.* **49**(73), 8081–8083 (2013). <https://doi.org/10.1039/C3CC45075A>
12. Leng, J.-D., Liu, J.-L., Tong, M.-L.: Unique nanoscale Cu<sub>II</sub>36Ln<sub>III</sub>24 (Ln = Dy and Gd) metallo-rings. *Chem. Commun.* **48**(43), 5286–5288 (2012). <https://doi.org/10.1039/C2CC30521F>
13. Podgornii, D., et al.: Heterometallic Fe<sub>18</sub>M<sub>6</sub> (M=Y, Gd, Dy) pivalate wheels display solventinduced polymorphism. *Cryst. Growth Des.* **22**(9), 5526–5534 (2022). <https://doi.org/10.1021/acs.cgd.2c00620>
14. Botezat, O., van Leusen, J., Kravtsov, V.C., Kögerler, P., Baca, S.G.: Ultralarge 3d/4f coordination wheels: from carboxylate/amino alcohol-supported Fe<sub>4</sub>Ln<sub>2</sub> to Fe<sub>18</sub>Ln<sub>6</sub> rings. *Inorg. Chem.* **56**(4), 1814–1822 (2017). <https://doi.org/10.1021/acs.inorgchem.6b02100>



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15. Gerbeleu, N.V., Batsanov, A.S., Timko, G.A., Struchkov, Y.T., Indrichan, K.M., Popovich, G.A.: Synthesis and structure of tri- and hexanuclear  $\mu_3$ -oxopivalates of iron(III). *Dokl. Akad. Nauk SSSR* **293**, 364–367 (1987)
16. Sheldrick, G.M.: A short history of SHELX. *Acta Cryst.* **A64**(1), 112–122 (2008).  
<https://doi.org/10.1107/S0108767307043930>
17. Sheldrick, G.M.: Crystal structure refinement with SHELXL. *Acta Cryst.* **C71**(1), 3–8 (2015).  
<https://doi.org/10.1107/S2053229614024218>
18. Spek, A.L.: PLATON SQUEEZE: a tool for the calculation of the disordered solvent contribution to the calculated structure factors. *Acta Crystallogr.* **C71**, 9–18 (2015).  
<https://doi.org/10.1107/S2053229614024929>
19. Brese, N.E., O’Keeffe, M.: Bond-valence parameters for solids. *Acta Crystallogr.* **B47**, 192–197 (1991). <https://doi.org/10.1107/S0108768190011041>
20. Spek, A.L.: Structure validation in chemical crystallography. *Acta. Crystallogr.* **65**(2), 148–155 (2009). <https://doi.org/10.1107/S090744490804362X>
21. Abourahma, H., Moulton, B., Kravtsov, V., Zaworotko, M.J.: Supramolecular isomerism in coordination compounds: nanoscale molecular hexagons and chains. *J. Am. Chem. Soc.* **124**, 9990–9991 (2002). <https://doi.org/10.1021/ja027371v>
22. Anghel, S., et al.: Site-selective luminescence spectroscopy of bound excitons and local band structure of chlorine intercalated 2H- and 3R-MoS<sub>2</sub> polytypes. *J. Lumin.* **177**, 331–336 (2016).  
<https://doi.org/10.1016/j.jlumin.2016.05.017>